PESTICIDE FORMULATION & RESIDUE ANALYTICAL CENTRE, PMD, NIPHM, HYDERABAD

Sr. No. in Scope NABL / NON NABL

Flow Chart for Analysis of Sulphur in formulation

Date of Analysis	

Sl. No.	Step	Executi	Executed By	
1.	Sample No.			
2.	Name of Sample			
2.1	Sample Description			
3.	Preparation of Standard solutions (0.1 N):	R ₁	$\mathbf{R_2}$	
3.1	Weigh 24.8 g of Sodium thiosulphate and dissolve in water to get 1 L solution	g	g	
3.2	Note the Serial No. in the balance logbook			
4.	Standardization of Sodium thiosulphate:			
4.1	Weigh 0.07-0.08 g of KIO₃ in a conical flask			
4.2	Note the Serial No. in the balance logbook			
4.3	Add 40-50 ml of water and dissolve			
4.4	Add 2 g of KI			
4.5	Note the Serial No. in the balance logbook			
4.6	Add 2 ml of 5N H ₂ SO ₄			
4.7	Titrate Immediately with Sodium thiosulphate			
	solution in burette till pale yellow appears			
4.8	Add 1 ml of starch indicator (Color of solution			
	changes to blue)			
4.9	Then again titrate with Na ₂ S ₂ O ₃ till it changes to			
	colorless (endpoint)			
4.10	Titre value is			
5.	Preparation of Iodine solution			
5.1	Weigh 30-40 g of KI into an Iodine flask	g	g	
5.2	Note the Serial No. in the balance logbook			
5.3	Add 50-60 ml of water and dissolve			
5.4	Add 13 g of Iodine to 5.3 and dissolve thoroughly	g	g	
5.5	Note the Serial No. in the balance logbook			
5.6	Transfer into a amber colored bottle (Keep it for overnight)			
6.	Standardization of Iodine solution			
6.1	Take 25 ml of standardized Iodine solution in a conical flask			
6.2	Add 1 ml of starch solution			

	ciue Foi	mulation &	Residue Analytical Ce	entro	e, PMD, NIPHM, Hyderabad
:	FC-PF	-245	Document Name		Flow chart for analysis of Sulphur in Formulation
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Revision Date : 26/03/2		2014	Next Revision Date		26/03/2016
Prepared By		Checked By			Approved &Issued By
Mrs. T. Sridevi (Assistant Scientific Officer)			Mr. C.V. Rao		Dr. Abhay Ekbote (Director PM & Quality Manager)
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6.3	Titrate against sodium thiosulphate solution, end	
	point will be colorless to blue	
6.4	Titre value is	
7.	Procedure	
7.1	Weigh accurately 0.1 g of sample (0.125 g of 80%	
	WP) and transfer into a 500 ml GG flat bottom flask	
7.2	Add 10 ml alcohol to dissolve sulphur (If the	
	formulation is DP)	
7.3	Add 30-40 ml of water	
7.4	Add 2 g Sodium sulphite	
7.5	Add liquid paraffin and few glass beads	
7.6	Fix the flak to the water cooled condenser & keep for	
	reflux for 40 minutes	
7.7	Cool the flask, rinse the condenser with water and	
	remove the flask	
7.8	Add 10 ml of formaldehyde	
7.9	Add 10 ml of Carbon tetrachloride	
7.10	Add 10 ml of Acetic acid	
7.11	Titrate immediately with standard Iodine solution	
	using freshly prepared starch solution as indicator	
7.12	End point will be the first appearance of blue color	
	throughout the solution	
7.13	Follow the same procedure from 7.2 to 7.12 for	
	reagent blank	
7.14	Titre value for sample is	
7.15	Titre value for blank is	

6. Calculation:

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Normality of Na ₂ S ₂ O ₃ X volume of Na ₂ S ₂ O ₃ taken
Normality of Iodine =
Volume of Iodine consumed

	(mL of Iodine for sample) - (mL of Iodine for blank) X N X 3.206
Active ingredient content, =	
% by mass	Weight of sample (g)

Where N = Normality of Iodine solution

Result:

	Nesuit.			
Sl. No.	Name of Test	Result	Unit	Method of Analysis
1.	Active ingredient (Sulfur)		%	IS: 6444 - 1979 (Reaffirmed 2002)
Referen	ce in daily workbook:			

	Name	
Analyzed by	Dated signature	
	Name	
Checked by	Dated signature	

Name of the Laboratory : P	esti	cide For	mulation &	Residue Analytical Co	entr	tre, PMD, NIPHM, Hyderabad
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